

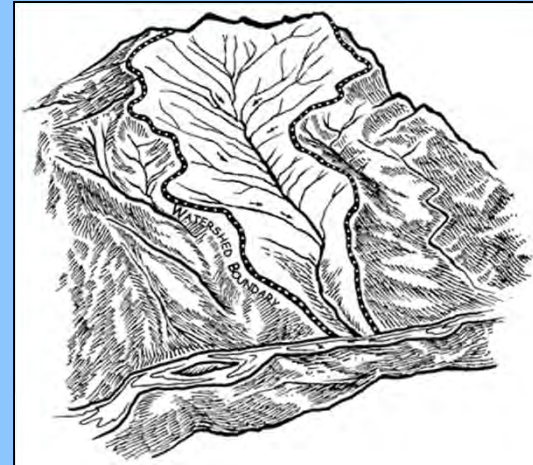
Citizens' Environmental
Monitoring Program
**Water Quality Monitor
Training**



Training Objectives

Day 1

What
Why




Days 2 & 3

How





CLEAN WATER  STRONG COMMUNITIES

- The concept of a “waterkeeper”
- Waterkeeper Alliance
- Cook Inletkeeper began in 1995

Vision:

Cook Inletkeeper works to
guarantee clean water for:

Abundant Fish and Wildlife
Strong Communities
Lasting Jobs
Renewable Energy
Public Property Rights

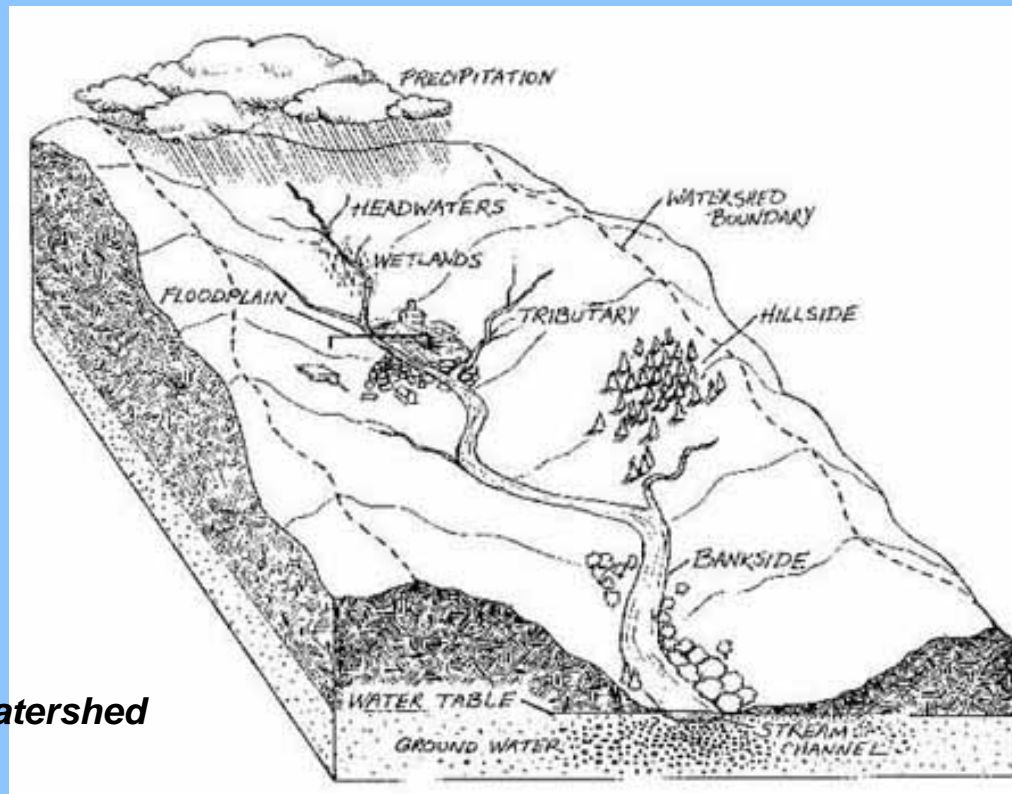
Mission:

To protect
Alaska’s Cook Inlet watershed
and
the life it sustains



Watersheds

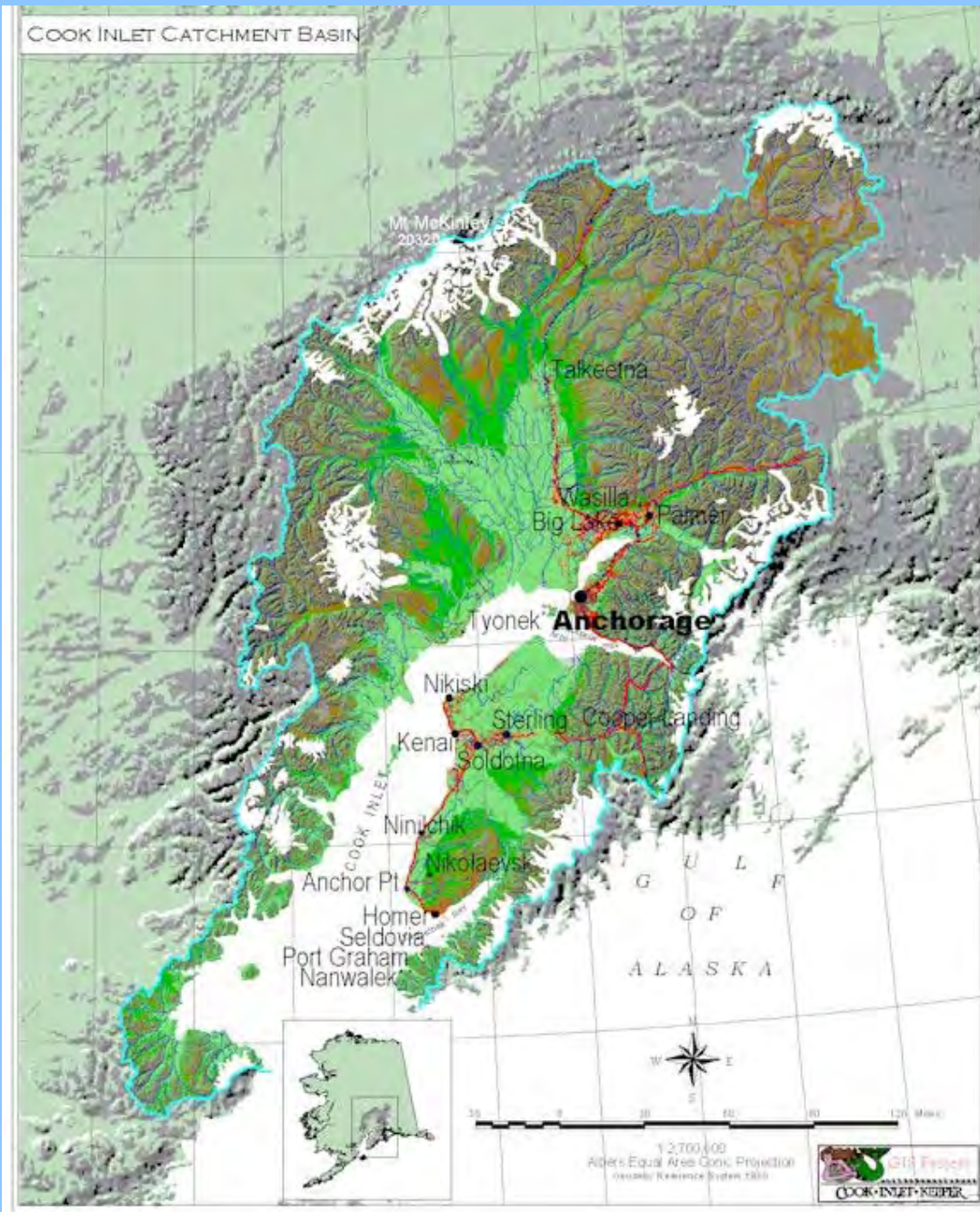
A watershed is the area of land from which runoff (from rain, snow, and springs) drains to a stream, river, lake, or other body of water. Its boundaries can be identified by locating the highest points of lands around the water body.



Cross section of a watershed

The Cook Inlet Watershed

The Cook Inlet watershed covers over 39,000 square miles of Southcentral Alaska. Melting snow and ice from mount McKinley, the Chugach Mountains and the Aleutian Range drains into rivers such as the Susitna, Matanuska and Kenai, which feed the productive waters of Cook Inlet.



CEMP: Citizens' Environmental Monitoring Program

- Goals and Objectives:
 - Inventory baseline water quality in the waters of Cook Inlet Basin
 - Detect and report significant changes and track water quality trends
 - Raise public awareness of the importance of water quality through hands on involvement

Citizen Involvement

Over 500 citizens trained



CEMP Partnership



**Homer Soil and Water
Conservation District**



**KENAI WATERSHED
FORUM**



**Wasilla Soil & Water
Conservation District**



**Upper Susitna Soil &
Water Conservation
District**



Resurrection Bay Conservation Alliance



CEMP Partnership

CEMP Monitoring Partners:

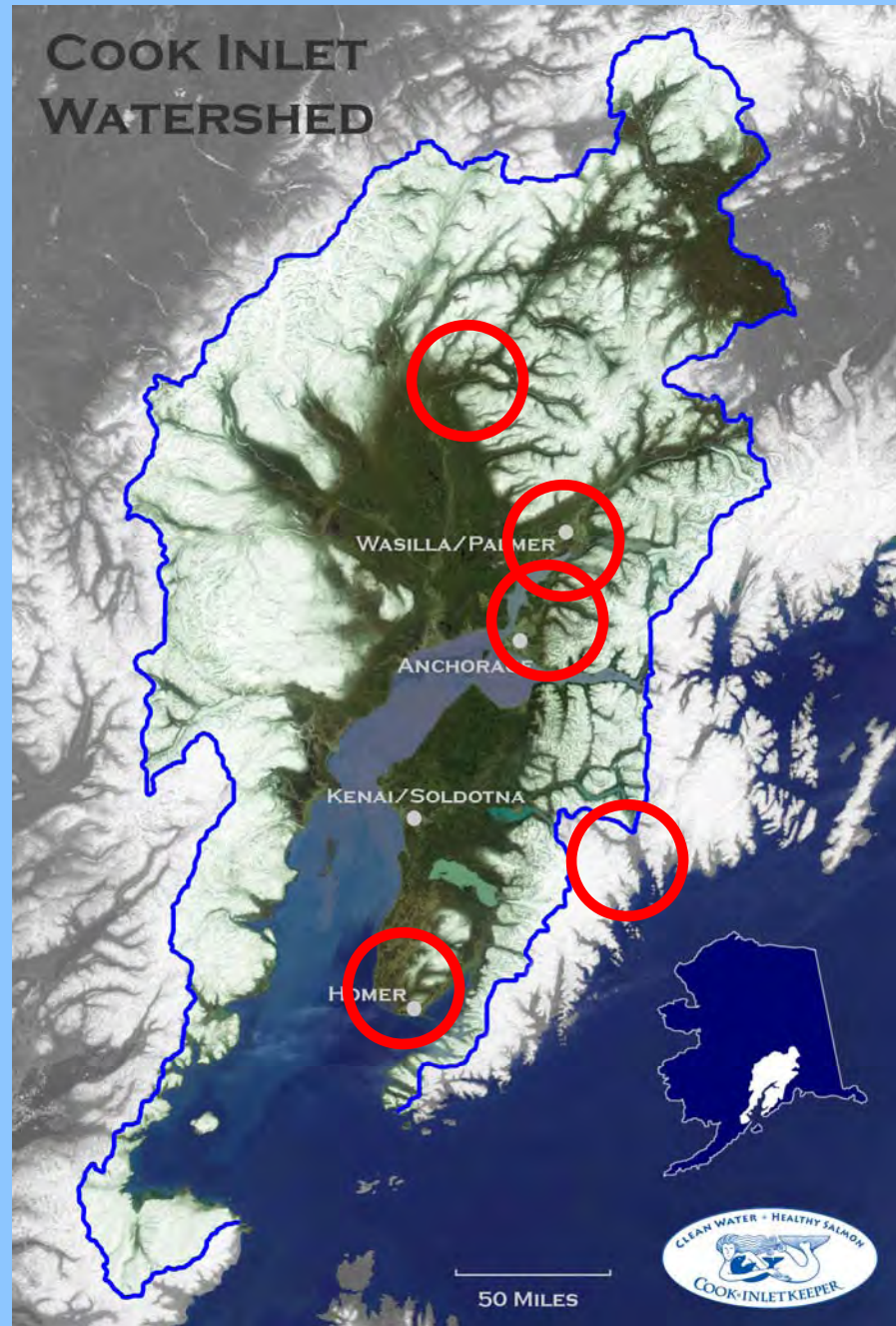
Resurrection Bay Conservation Alliance, Anchorage Waterways Council, Mat-Su Lake Monitoring, Wasilla SWCD, Upper Susitna SWCD

CEMP Outreach Partners:

Homer SWCD, Kenai Watershed Forum

CEMP Technical Partners:

UAA Environment and Natural Resources Institute (ENRI)



Baseline Data: A long term success story!

Baseline Reports

Filling in the gaps:

- Water quality monitoring
- Bioassessments
- Habitat Assessments
- GIS Analysis
- Invasive plant surveys
- Temperature monitoring

2010 CEMP Strategic Monitoring Plan (based off of data through 2009)	GOAL	Bridge Creek	Ruby Creek	Two Moose Creek	Beaver Creek	U T
Water Quality						
Years Monitored	>5	10	13	7	8	
Site Visits	>80	61	74	40	107	
Years w/ >12 visits	5	2	2	0	7	
Months w/ > 5 visits	12	5	10	6	11	
Number of summer visits	40	33	34	28	58	
enough WQ data for baseline?		no	no	no	yes	
Standard Deviation						
Annual temperature	4.60	4.62	3.75	4.15	5.08	
Summer temperature (June-Aug)	2.00	2.84	2.49	1.79	3.16	
Dissolved Oxygen (mg/L)	0.10	0.05	0.08	0.11	0.11	
pH (Hanna)	0.49	0.36	0.37	0.43	0.38	
good enough for trend analysis?		n/a	n/a	n/a	yes	
CEMP Priority						
High, Medium, Low		Medium	High	High	High	M
continue monitoring in 2010?		yes	yes	yes	no	
Temperature Monitoring						
Years monitored (May-Oct)	3	0	5	5	1	
Start Year		no	yes	yes		
Close-out Year						
Bioassessment						
Number of visits	6	0	13	11	12	
Begin Sampling						
End Sampling						
Habitat Analysis Year						
GIS Analysis Year						
Trend Analysis			X	X	X	
Baseline		x				x
Volunteer(s)		Brown	Milburn	York	Wagners	L
Other Relevant Sites (within watershed)		KB-490				
Proposed Baseline Report Year		2013	2013	2013	2010	
Re-assessment Year						

Bioassessment = BUGS!



We monitor 5 sites twice a year – once in June and again in August!

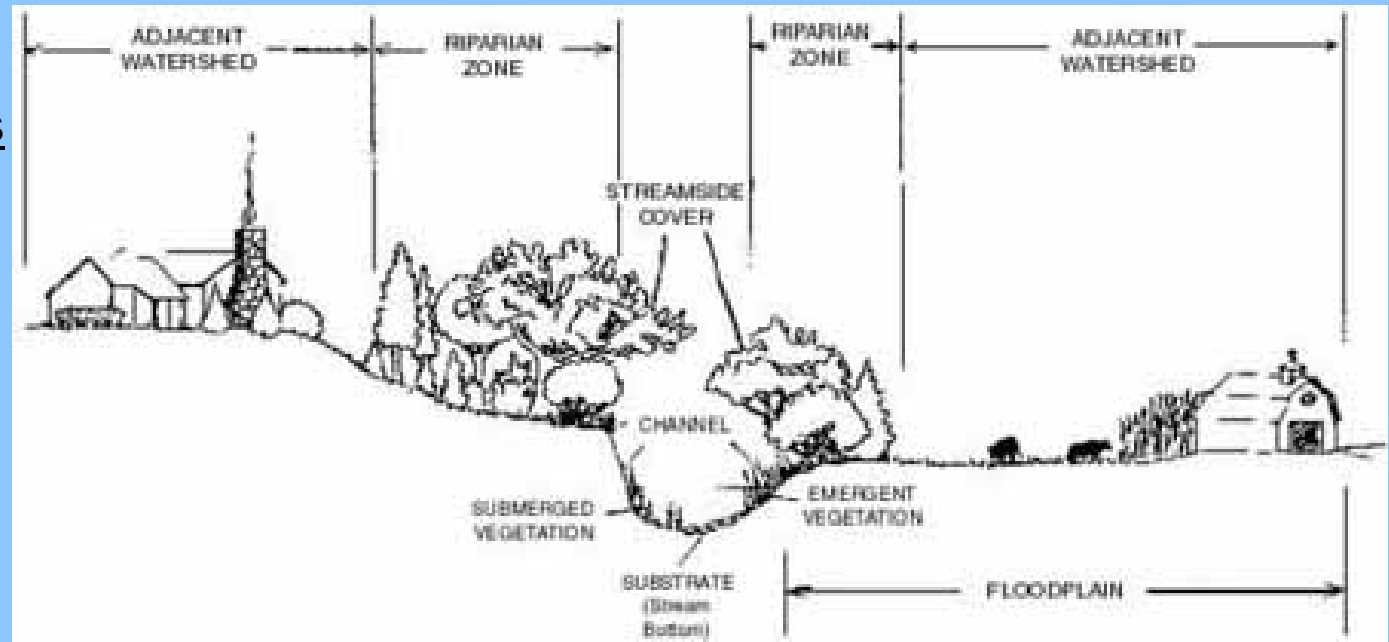


The Living Stream Environment

Components of the stream system

A healthy stream is a busy place.

- Wildlife and birds
- Vegetation
- Fish
- Insects and other macro invertebrates



Human activities shape and alter many of these stream characteristics

Homes, farms, other developments

Septic systems/outhouses, rural runoff, increase in impervious surfaces

Towns and cities – urban runoff

Logging, mining, gravel extraction activities

CEMP Monitoring Sites

Anchor River

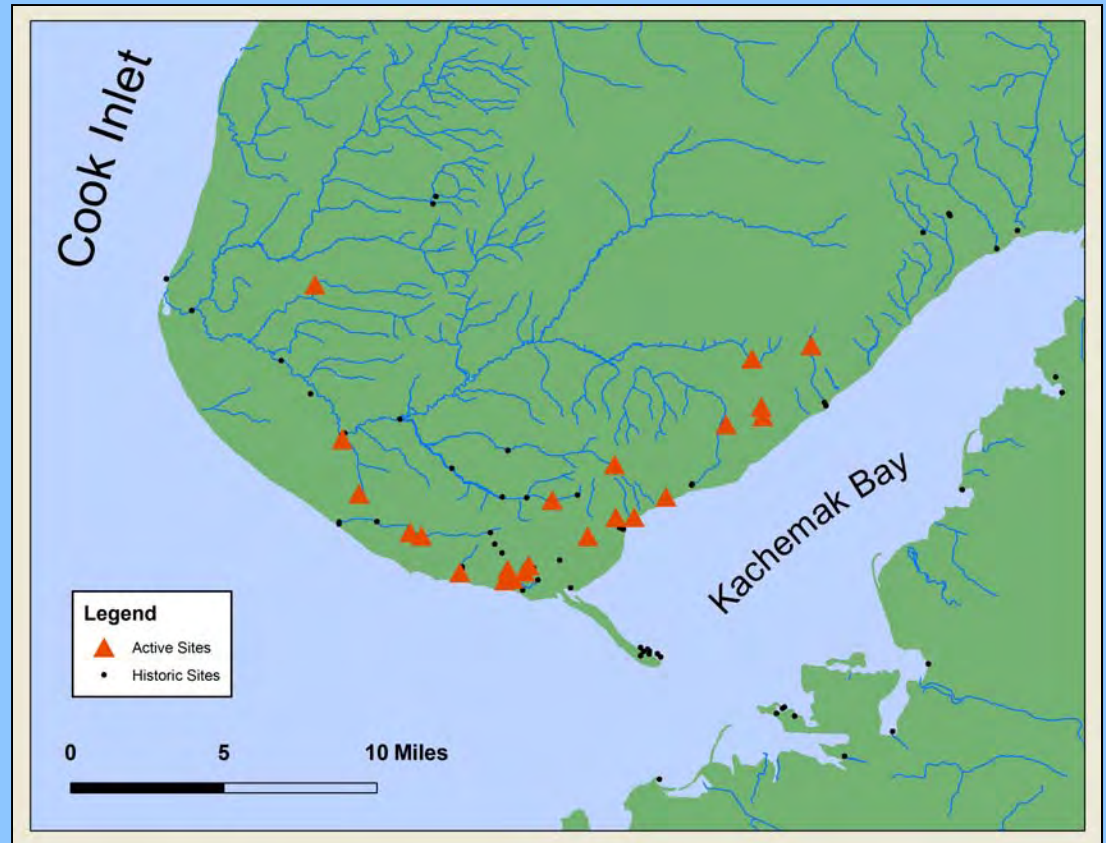
watershed:

Beaver Creek Ruby Creek
Two Moose
Bridge Creek

Kachemak Bay

watershed:

Woodard Creek	<i>Bidarka Creek</i>
Mariner Creek	Palmer Creek/Beluga Slough
Fritz Creek	<i>McNeil Canyon</i>
Rice Creek	Miller Creek
Diamond Creek	



Site Selection



- **Representative**
- **Safely and reasonably accessible**
- **Prioritization based on uses, database needs, and potential threats**

Two Moose Creek



Ruby Creek



“No Name” Creek



Beaver Creek



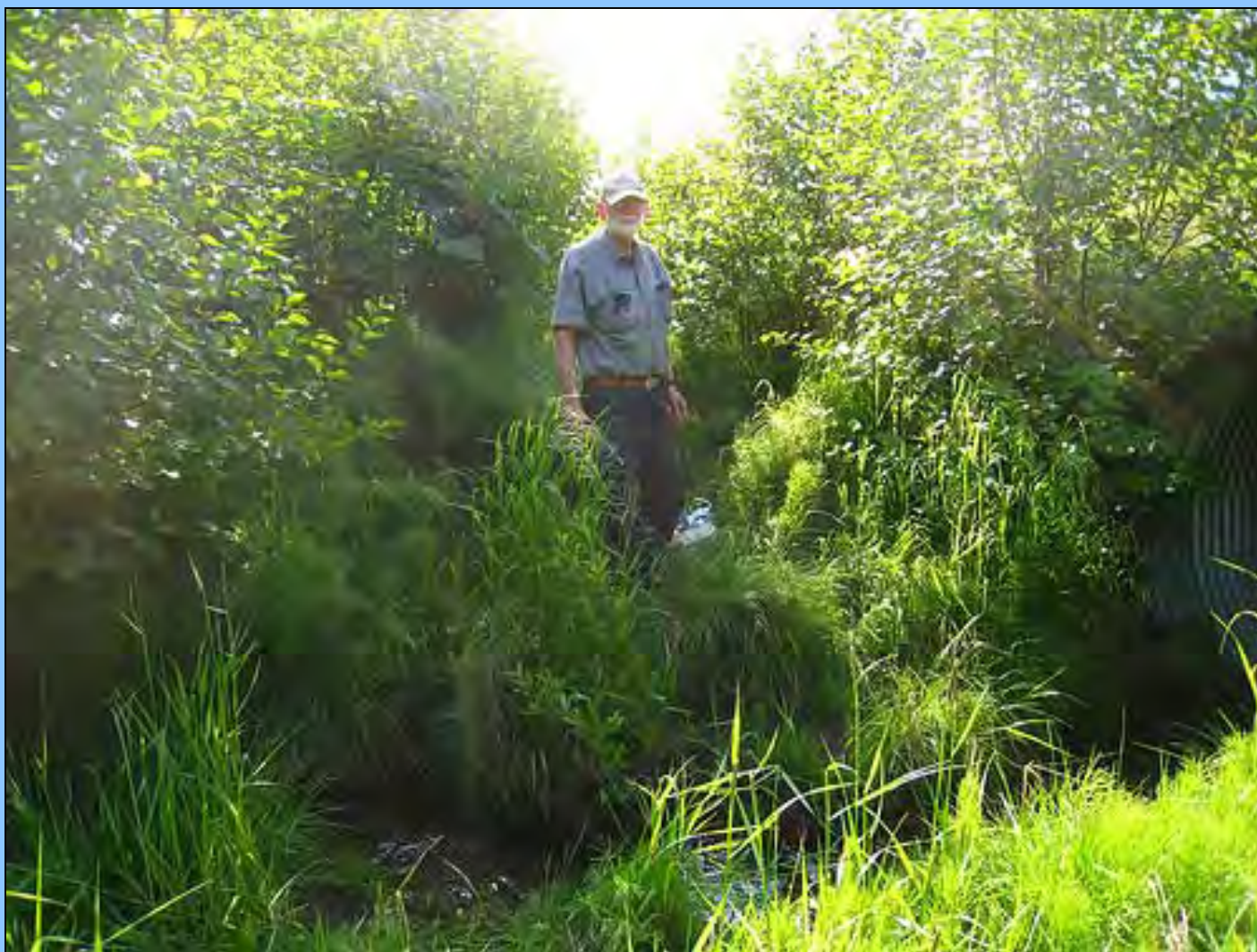
Bridge Creek



Lower and Upper Diamond Creek



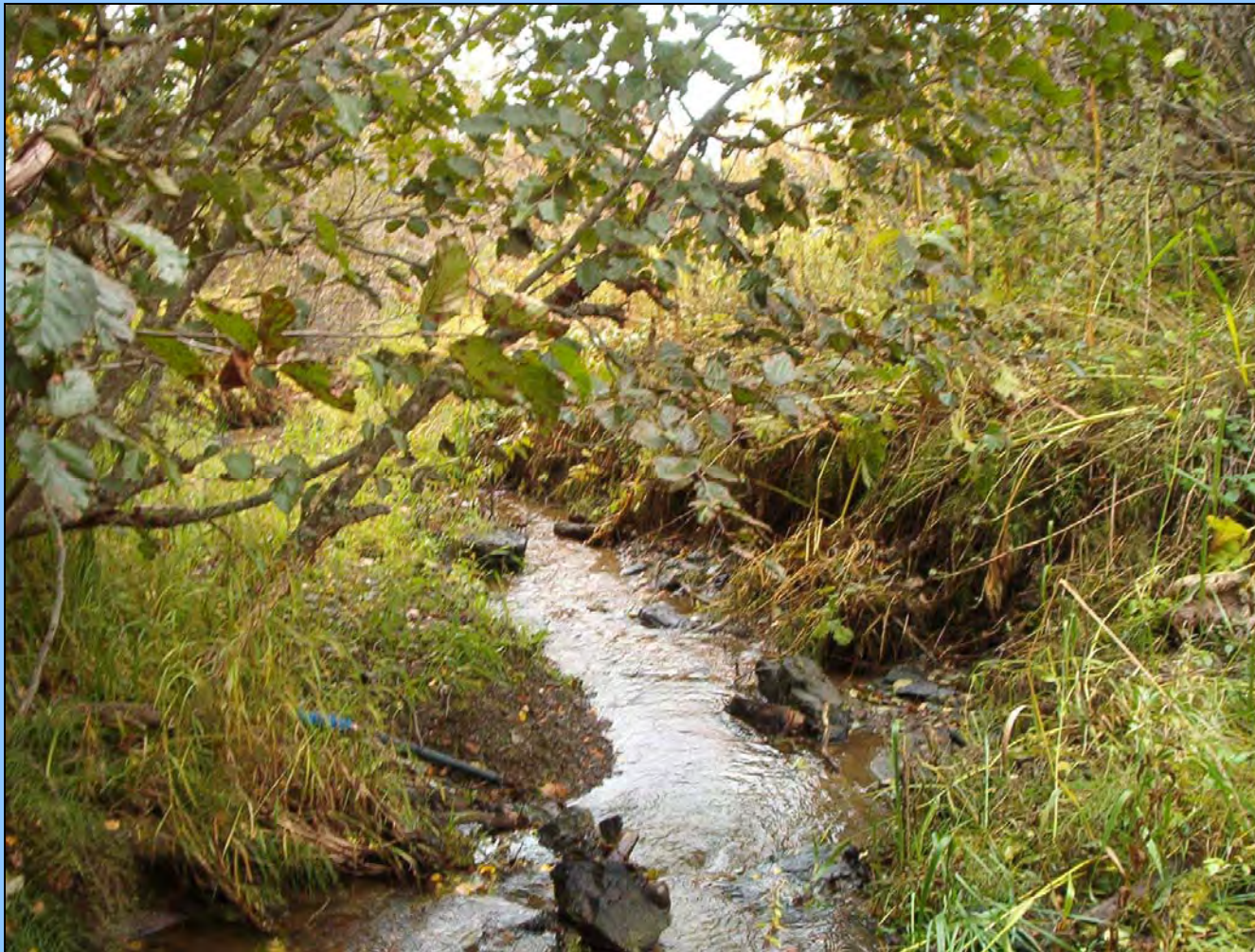
Bidarka Creek



Upper and Lower Woodard Creek



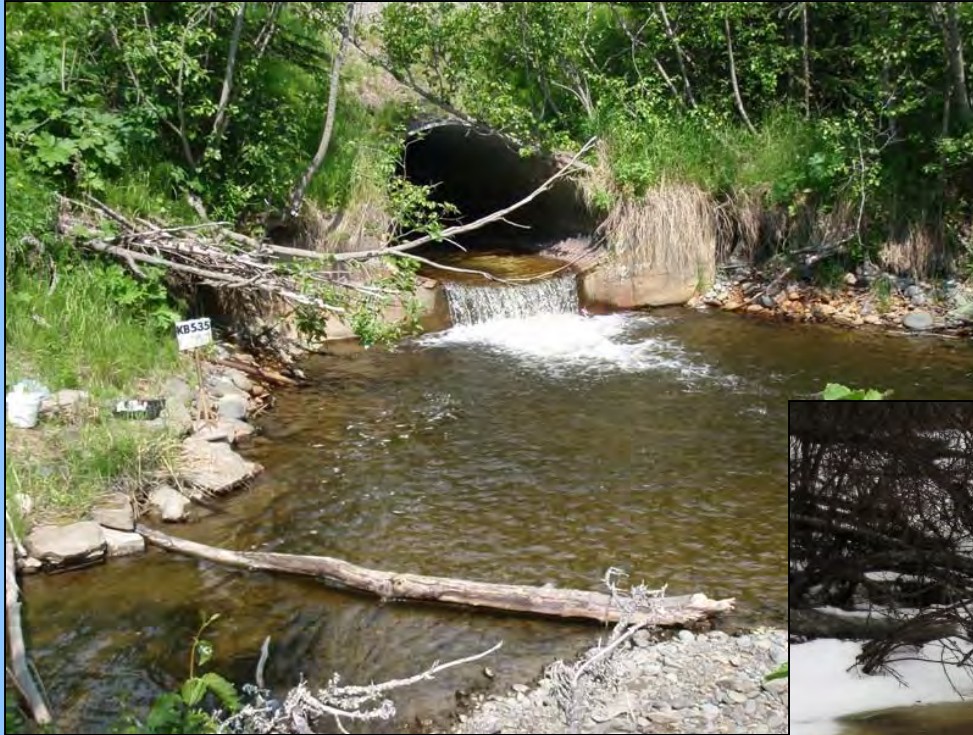
Palmer Creek



Miller Creek



Upper and Lower Fritz Creek



Rice Creek



McNeil Creek



What is Water Quality?

Pollution is broadly divided into two classes according to its source:

Point source

Nonpoint



Common sources of pollution to streams include:

Agricultural

Municipal dischargers

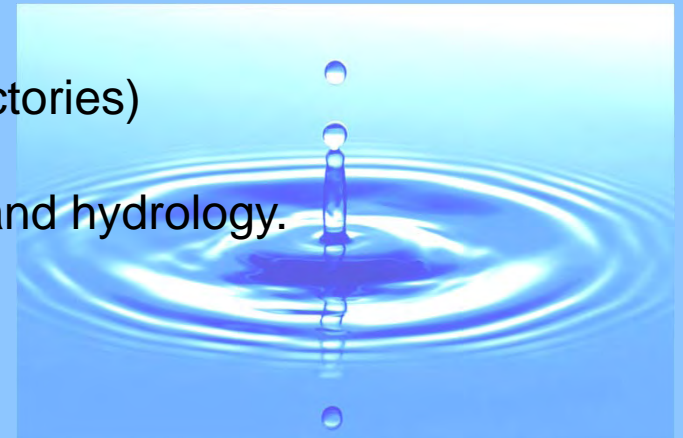
Urban runoff

Mining

industrial dischargers (factories)

forestry activities

modifications to stream habitat and hydrology.



What We Monitor

- **Chemical**
- **Biological**
- **Physical**



WHY DO WE MONITOR WATER QUALITY?



The Need for Monitoring

- The federal Clean Water Act of 1972
- The single largest factor limiting our ability to make intelligent policy decisions is not having sufficient information (i.e. hard data).
- Growth in population and its associated pressures on water quality and natural resources
- The federal and state agencies charged with monitoring and protecting water quality have found it increasingly difficult to fulfill their mandates.

Water quality monitoring can be used for many purposes:

To identify whether waters are meeting designated uses.

To identify specific pollutants and sources of pollution.

To determine trends.

To screen for impairment.

<u>Source</u>	<u>Common Associated Indicators</u>
Cropland	Turbidity , phosphorus, nitrates, temperature , conductivity
Forestry harvest	Turbidity , temperature , conductivity
Grazing land	Fecal bacteria , turbidity , phosphorus, nitrates, temperature
Industrial discharge	Temperature , conductivity , toxics, pH
Mining	pH , alkalinity, conductivity
Septic systems	Fecal bacteria , nitrates, phosphorus, dissolved oxygen / biochemical oxygen demand, conductivity , temperature
Sewage treatment plants	Dissolved oxygen and biochemical oxygen demand, turbidity , conductivity , phosphorus, nitrates, fecal bacteria , temperature , pH
Construction	Turbidity , temperature , dissolved oxygen and biochemical oxygen demand, conductivity , and toxics
Urban runoff	Turbidity , phosphorus, nitrates, temperature , conductivity , dissolved oxygen and biochemical oxygen demand

Some Types of Water Quality Monitoring

- **Baseline Monitoring**
- **Compliance and Enforcement Monitoring**

Sampling Schedule

- Samples are to be taken on the second and last Sundays May - August and on the last Sunday September - April for a total of 16 x per year.
- 2:00 PM.
- The quality of the data collected by our program depends on regular and consistent monitoring

Safety

The following are some basic common sense safety rules. At the site:

Always let someone else know where you are, when you intend to return

Complete a volunteer form.

If you drive, park in a safe location. Be sure your car doesn't pose a hazard to other drivers.

Never cross private property without the permission of the landowner.

Watch for wildlife and insects. Know what to do if you get bitten or stung.

Never drink the water in a stream. After monitoring, wash your hands with antibacterial soap.

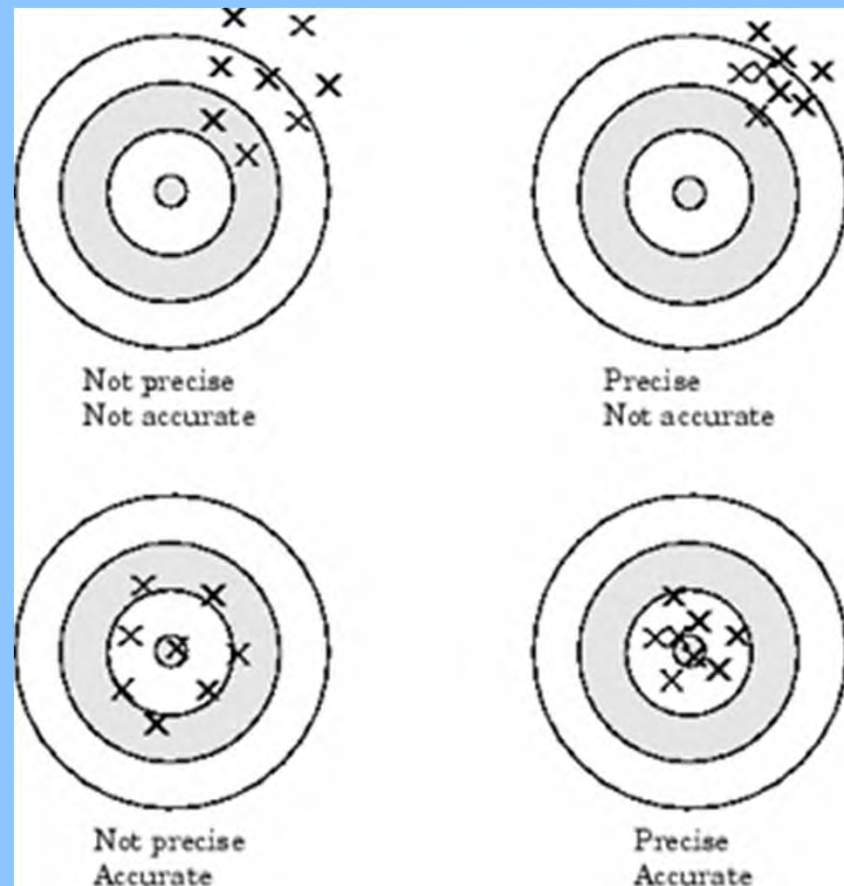
If the water appears to be severely polluted, contact the Monitoring Coordinator.

Disturb streamside vegetation as little as possible.

If at any time you feel uncomfortable about the condition of the stream or your surroundings, stop monitoring and leave the site at once. Your safety is more important than the data!

Quality Control & Assurance

- Precision & Accuracy
- Training & Recertification
 - DQO
- Site visits & Split samples
- Data Management



(www.mathworks.com)

Quality Assurance Plans

- Quality Management Plan
 - Guiding Umbrella Document for CEMP Partnership
 - Standard Operating Procedures (SOPs)
- Quality Assurance Project Plan(s) (QAPP)
 - Field Procedures Manual
 - Standard Operating Procedures (SOPs)
- Data Management
 - Access Database
 - Quarterly newsletters & Annual report

What and Why

Field Observations

- Air Temperature
- Wind & Weather
- Water Surface & Substrate
- Comments & Observations
- Photos & Sketches

Water Quality Parameters

- Color*
- Turbidity (Clarity)
- Water Temperature
- pH
- Conductivity
- Dissolved Oxygen
- Coliform Bacteria

Recording your data

Kachemak Bay & Anchor River CEMP Water Quality Data Sheet <small>Revised 3/2009</small>				Office Use Only		All Chemicals Current? <input type="checkbox"/>	
				Entry Date	Edit		
Sample Information							
Site ID				Monitoring Kit Number			
Date				Kit Condition			
Time				Comments			
Volunteer Information							
	Print Name	Signature		Mileage	Hours		
Volunteer 1							
Volunteer 2							
Volunteer 3							
Volunteer 4							
Hanna Meter Calibration				Weather		Wind	
Meter #	Date			Clear	Mph		Direction
	Temp	Cal?		Partly Cloudy	<1	N	
pH 7 (Initial)		Y/N		Cloudy	1-3	NE	
pH 4 (Final)		Y/N		Precipitation	4-7	E	
Cond(Initial)		X		Fog or Haze	8-12	SE	
Cond(Final)		Y/N			13-18	S	
Precipitation				Type (circle one)		Sample Location	
Last 24 hr. (inches)	Rain	Hail		Number of Days Similar	19-24	SW	
	Snow	Sleet		Temp, °F	25-31	W	
					32-38	NW	
					39-46		
					47-54	Character	
					55-63	Calm	
					64-72	Steady	
					73 or greater	Variable	
					Gusting	Description	
						Pool	
						Riffle	
						Other	
Comments				Sketch			
Photos							
Photo #	Description			Camera #			
#							
#							
#							
				Additional Photos			

Return to: Cook Inletkeeper 3734 Ben Walters Lane Homer, AK 99603 907.235.4068				Date	Page 2 of 2	
				Site ID		
Water Temperature				Color		
Repeat if not within $\pm 0.5^{\circ}\text{C}$				2.5 Gallon 50 ml		
Replicate 1	Replicate 2	Replicate 3	Location (circle one)	Apparent Color		
Time			stream	BCS #		
Temp °C			bucket	2nd BCS#		
Take replicates 5 minutes apart				Record 2nd BCS if first one isn't an exact match.		
Turbidity Sample Collection				Colorimetric pH		
Bottle #	Time	Date	Location	Replicate 1 Replicate 2		
			stream bucket			
				Record to nearest 0.25 pH units		
				DQO MET (0.25)		
Hanna Meter				Wait for Hanna meter to stabilize before recording measurement. Repeat if any two of three replicates are not within: Conductivity $\pm 2\mu\text{S}$, pH ± 0.02 units		
Meter #				Replicate 1	Replicate 2	Replicate 3
Start Temp °C				pH		
Stop Temp °C				Conductivity		
Comments:				pH ± 0.02 units Conductivity $\pm 2\mu\text{S}$		
Dissolved Oxygen				Repeat if three replicates are not within 0.6 mg/L		
Fix Time				DQO MET:		
Fix Temperature °C				Replicate 1a	Replicate 1b	
Titration Date				Replicate 2a	Replicate 2b	
Titration Time				Replicate 3a	Replicate 3b	
Comments:						
Coliform Bacteria				1 ml 5 ml		
Time mixed		Location:	Date counted	E. coli Colonies		
Time plated		stream	Time counted	Total Coliform		
Easygel Exp.		bucket		Total Colonies		
Turbidity <small>for office use</small>						
Final Turbidity: 1. NTU 2. NTU						

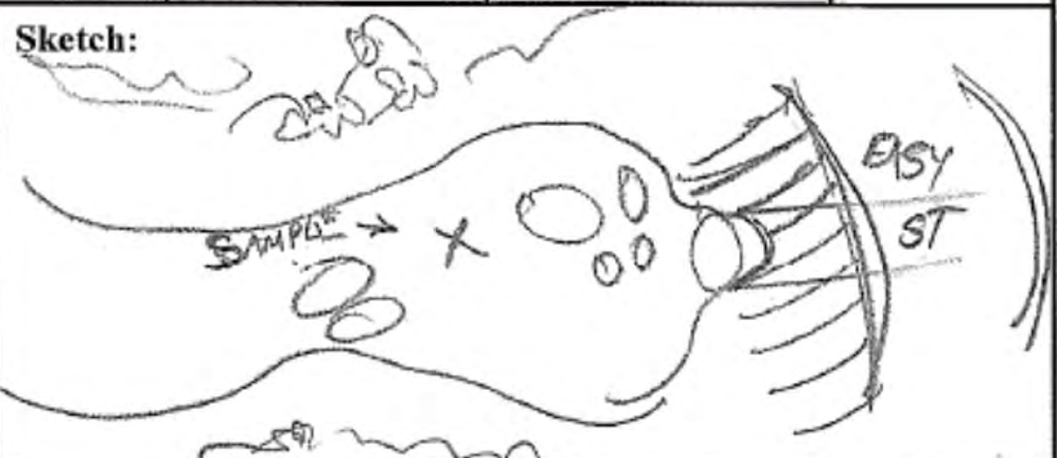
Field Observations

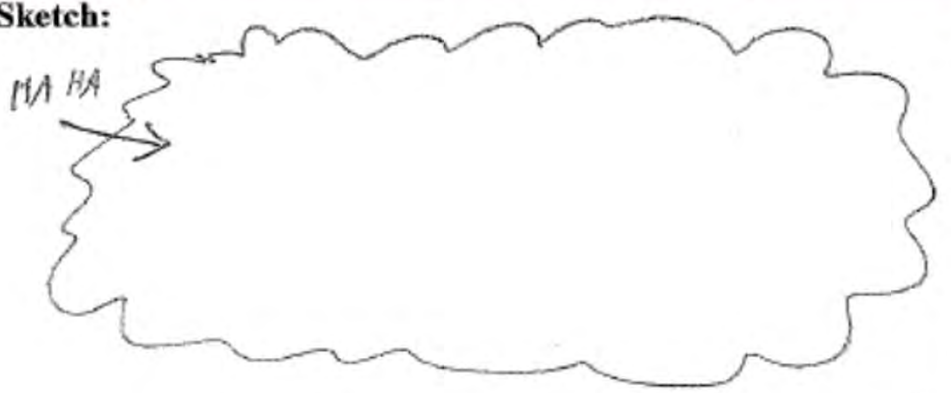
- Air Temperature
- Wind & Weather
- Water Surface & Substrate

Weather		Wind		Sample Location	
Clear		<u>Mph</u>	<u>Direction</u>	<u>Depth</u>	<u>Bottom</u>
Partly Cloudy		<1	N	0-6"	Silty
Cloudy		1-3	NE		
Precipitation		4-7	E	6-12"	Sandy
Fog or Haze		8-12	SE	12-36"	Muddy
		13-18	S		Gravel
		19-24	SW		
		25-31	W		
		32-38	NW		
		39-46			
Number of Days Similar		47-54	<u>Character</u>	<u>Description</u>	
		55-63	Calm	Pool	Riffle
		64-72	Steady		
		73 or greater	Variable		
Temp. °F			Gusting	Other:	

Precipitation		Type (circle one)	
Last 24 hr. (inches)		Rain	Hail
		Snow	Sleet

Comments & Sketches...

Comments: <u>STREAM LOW</u> 	Sketch: 
---	--

Comments: <u>3 FEET OF SNOW</u> <u>ON FROZEN CREEK</u> <u>CAN'T HEAR</u> <u>RUNNING WATER</u> 	Sketch: 
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Site Photos

- New system this year – quarterly photos
- Inletkeeper has a Flickr site for storage



Water Quality Parameters

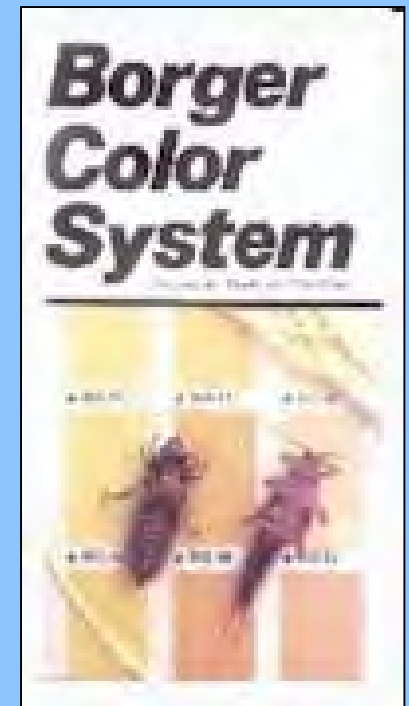
- Color
- Turbidity (Clarity)
- Water Temperature
- pH
- Conductivity
- Dissolved Oxygen
- Coliform Bacteria

Color

- Apparent color results from dissolved substances and suspended matter
- General but useful!
- CEMP utilizes your descriptive observations of apparent color as well as the BCS#(s)

Color	2.5 Gallon	50 ml
Apparent Color		
BCS #		
2nd BCS#		

Record 2nd BCS if first one isn't an exact match.

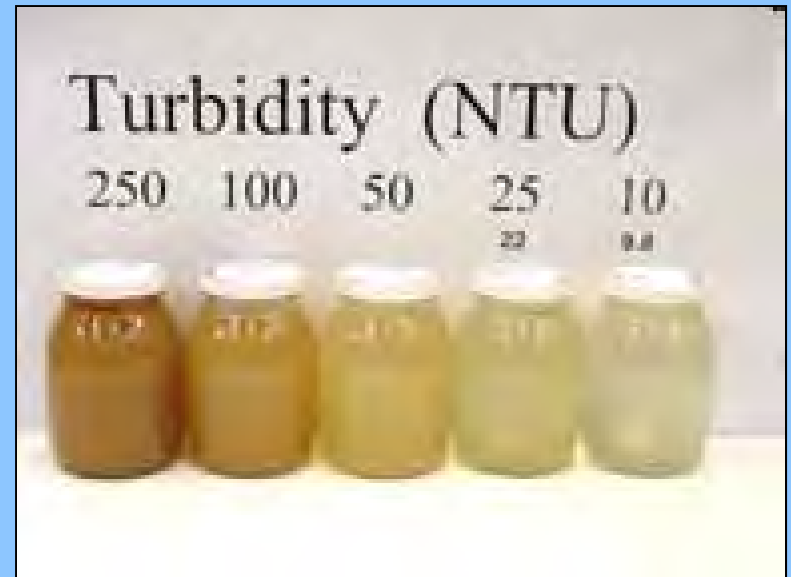


Turbidity (clarity)

- Turbidity is a measure of how much material suspended in the water decreases the passage of light through it
- Many affects on overall stream health

Turbidity Sample Collection			
Bottle #	Time	Date	Location
			stream bucket

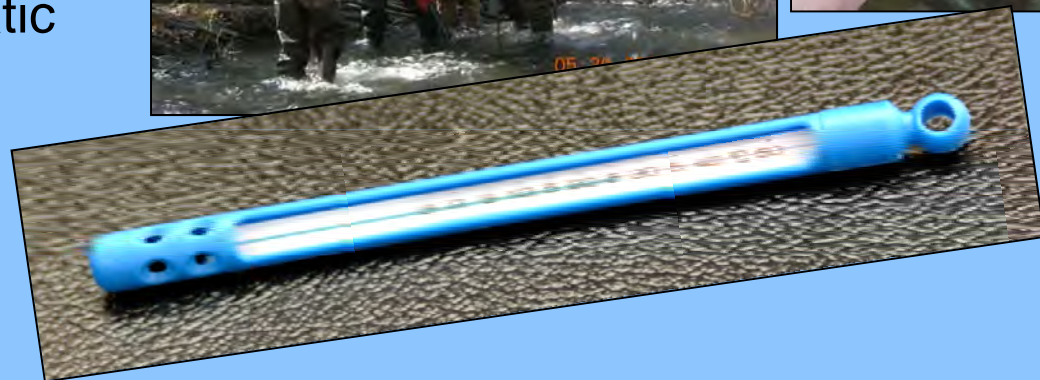
Turbidity <i>for office use</i>	
Final Turbidity:	1. _____ NTU 2. _____ NTU



Water Temperature

- Many factors influence water temperature
 - Air temperature
 - Stream flow
 - Riparian vegetation
 - Human inputs
- Affects
 - Rate of photosynthesis
 - Organisms metabolic processes
 - Distribution of aquatic life
 - Dissolved oxygen

Water Temperature			Repeat if not within $\pm 0.5^{\circ}\text{C}$	
	Replicate 1	Replicate 2	Replicate 3	Location (circle one)
Time				stream
Temp $^{\circ}\text{C}$				bucket
Take replicates 5 minutes apart			DQO MET	



pH

- A measure of how acidic or basic/alkaline a solution is
- pH ranges from 0 (acidic) to 14 (basic) and is on a *logarithmic* scale
- 6 – 8 is the pH range of most streams
- Changes in pH may be from:
 - Leaching of bedrock/soils
 - Human discharges
 - Aerosols, dusts, gasses from the air
 - Plant photosynthesis

Wait for Hanna meter to stabilize before recording measurement.
Repeat if any two of three replicates are not within: Conductivity $\pm 2\mu\text{S}$, pH ± 0.02 units

	Replicate 1	Replicate 2	Replicate 3	DQO MET
pH				

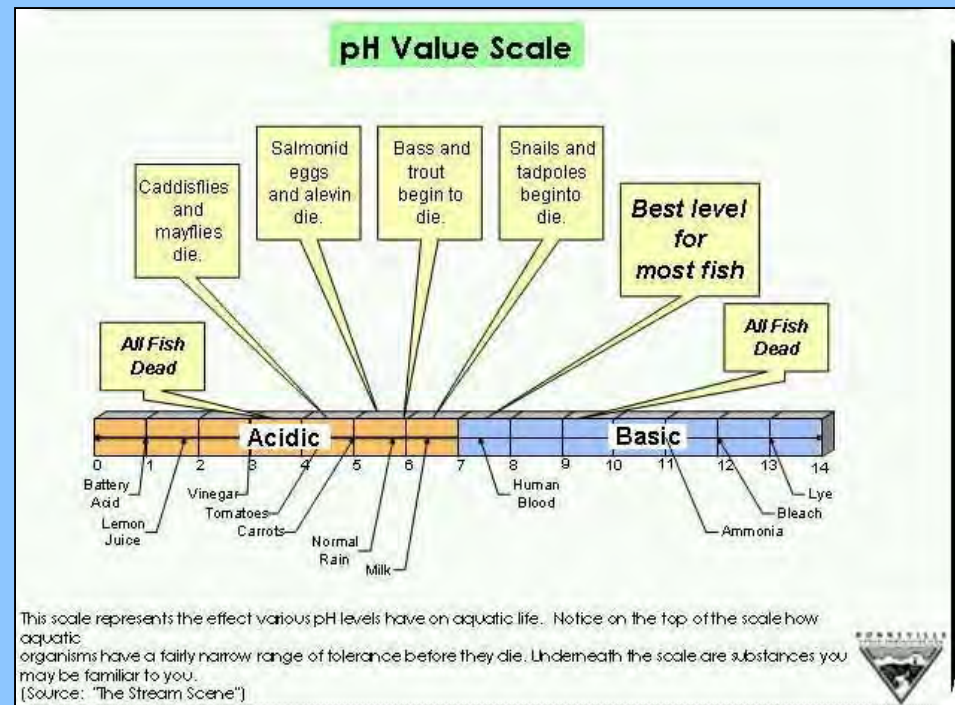
Colorimetric pH

Replicate 1 Replicate 2

--	--

Record to nearest 0.25 pH units

DQO MET (0.25)



Conductivity

- The conductance of water ($\mu\text{S}/\text{cm}$) = total dissolved solids
- Chloride, nitrate, sulfate, phosphate, sodium, magnesium, calcium, iron, aluminum
- Geology & human inputs will change conductivity

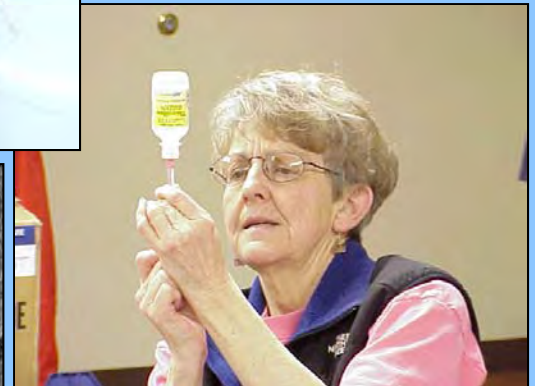
Hanna Meter				DQO MET
Conductivity				
				pH ± 0.02 units Conductivity $\pm 2\mu\text{S}$



Dissolved Oxygen

- One of the most important indicators!
- Concentration (mg/l) of oxygen dissolved in the water
- Saturation & biological oxygen demand
- Influences on DO:
 - Temperature
 - Wind & waves
 - Aquatic plants
 - Urban, agricultural, organic waste discharges
 - Dams

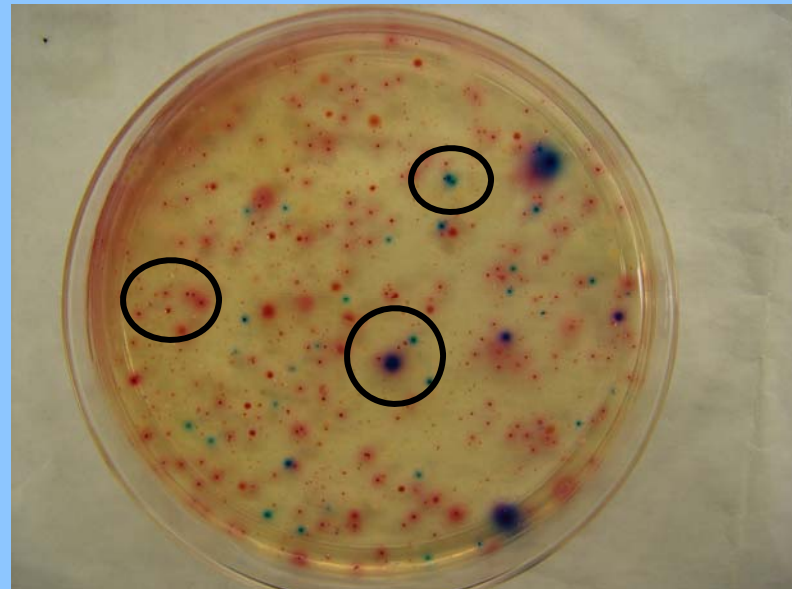
Dissolved Oxygen		Repeat if three	
Fix Time			
Fix Temperature °C			
Titration Date			
Titration Time			
		ithin 0.6 mg/L	DQO MET:
		ate 1a	Replicate 1b
		ate 2a	Replicate 2b
		Replicate 3a	Replicate 3b
nments:			



Coliform Bacteria

- *Indicator* bacteria of sewage pollution
 - Fecal coliform
 - *E. coli*
- Easier & far cheaper to monitor than pathogenic bacteria
- Urbanization, development, and increased population can all increase bacteria levels

Coliform Bacteria	
Time mixed	
Time plated	
Easygel Exp.	
Location: stream bucket	
Date counted	
Time counted	
	1 ml
<i>E. coli</i> Colonies	
Total Coliform	
Teal Colonies	



Volunteer Monitor Responsibilities

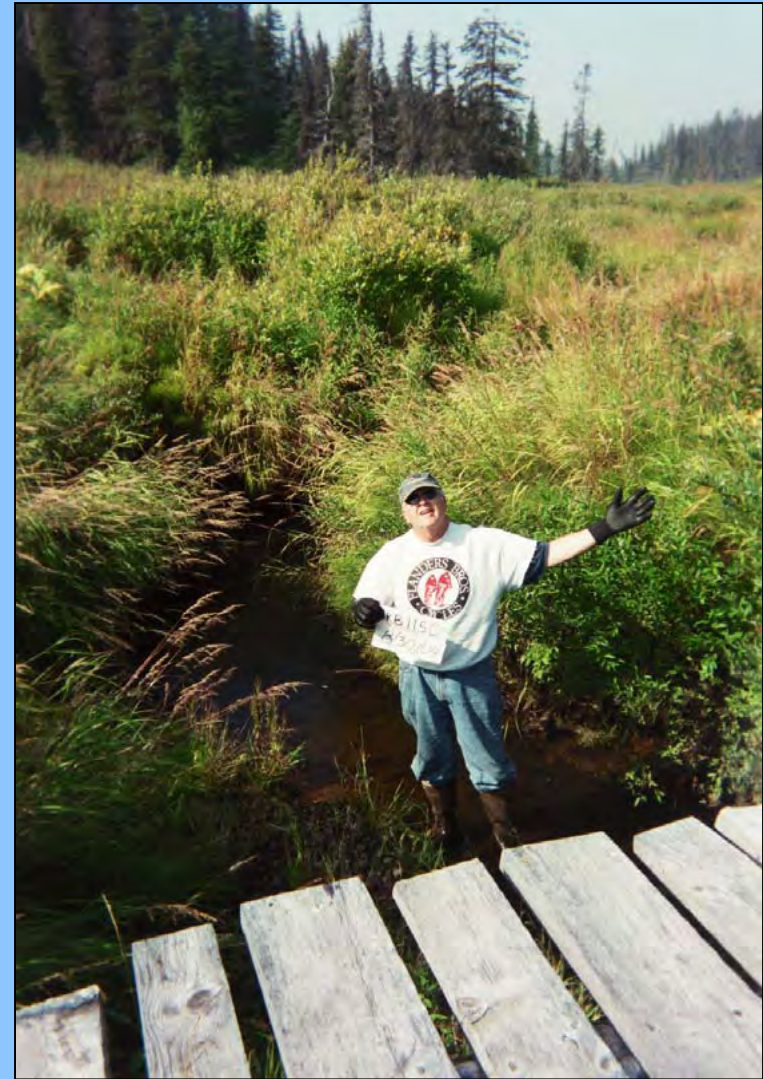
TRAINING

SAFETY

**MONITORING
SCHEDULE**

FIELD PROCEDURES

DATA MANAGEMENT



Volunteer Monitor Responsibilities (Cont.)

RESPONSE

KIT MANAGEMENT

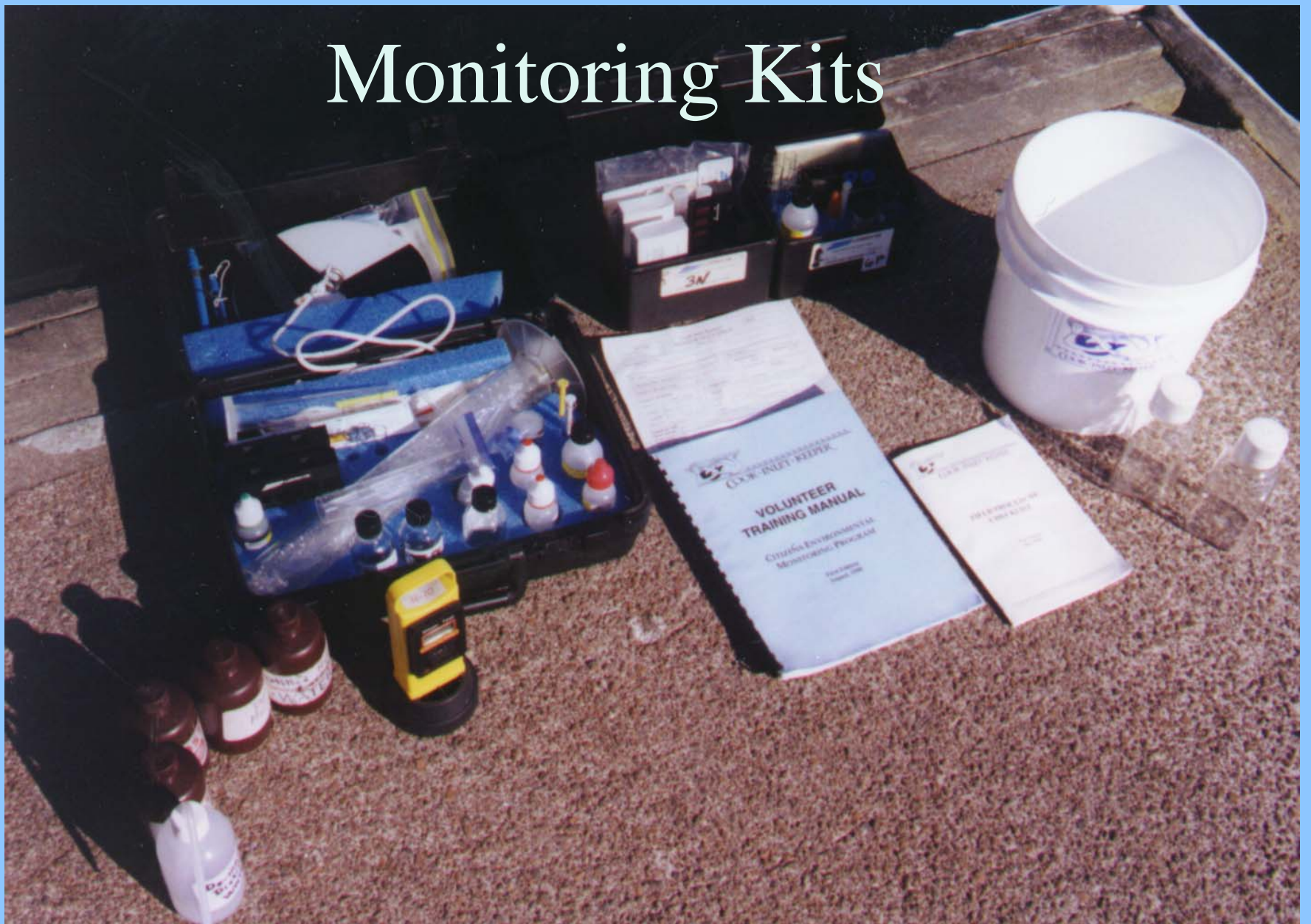
**CHEMICAL WASTE
MANAGEMENT**

**PERFORMANCE
STANDARD**

REPLICATE ANALYSIS



Monitoring Kits



When using chemicals:



Know your equipment, sampling instructions, and procedures before going out into the field.

Prepare labels and clean equipment before you get started.

Keep all equipment and chemicals away from small children. Many of the chemicals used in monitoring are poisonous.

Avoid contact between chemical reagents and skin, eye, nose, and mouth.

Never use your fingers to stopper a sample bottle (e.g., when you are shaking a solution).



Wear safety goggles when performing any chemical test or handling preservatives.

Know chemical cleanup and disposal procedures.

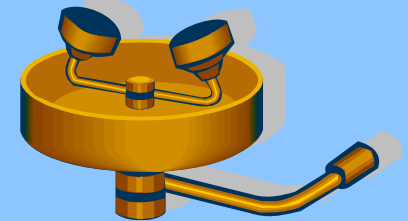
Wipe up all spills when they occur.

Return all unused chemicals to your program coordinator for safe disposal.

Close all containers tightly after use. Do not switch caps.

Know how to use and store chemicals.

Do not expose chemicals or equipment to temperature extremes or longterm direct sunshine.



Basic Equipment for Field Sampling

Listed below is some basic equipment appropriate for any volunteer field activity.

- Boots or waders
- “Field Clothes” - long sleeves and pants are best
- Rubber gloves
- Insect repellent/sunscreen
- Drinking water
- Clipboard
- Several pencils
- Tape measure
- Thermometer
- Field data sheet
- Camera and film, to document particular conditions



